

A Novel Synthesis of an A-Ring Precursor to 1 α -Hydroxyvitamin D

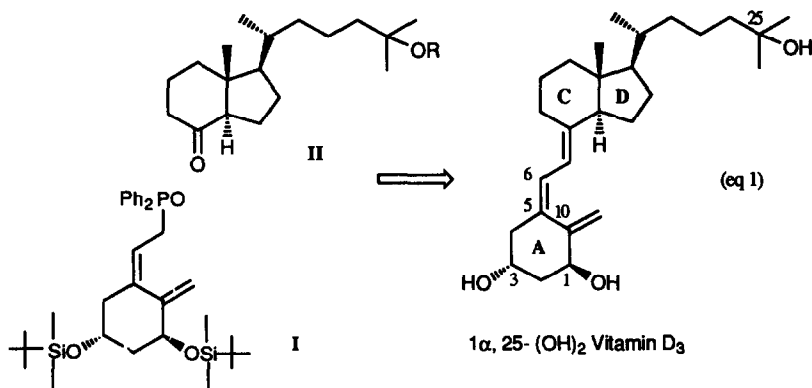
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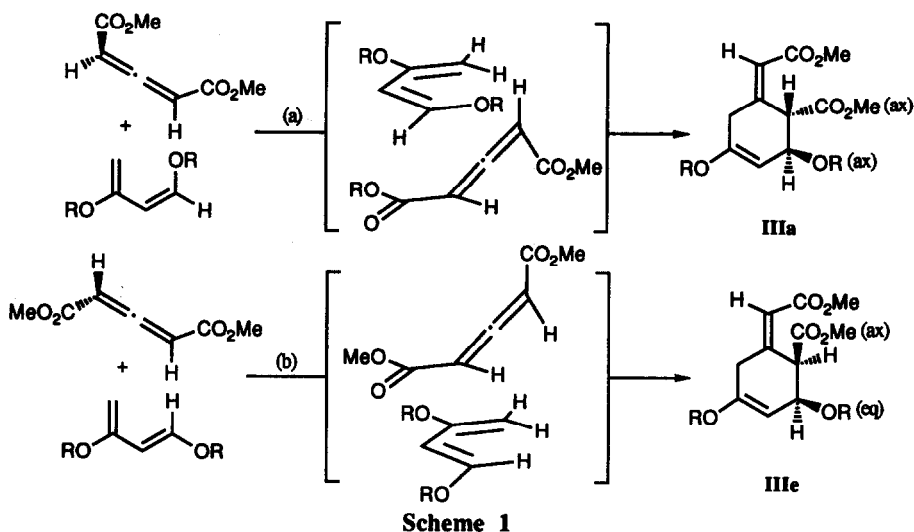
Abstract : A novel synthesis of the known, 1 α -hydroxyvitamin D A-ring precursor **10**, in racemic form, is described based upon (i) the stereoselective cycloaddition of allene-dienophile **2** onto 3-trimethylsilyloxyfuran **1** to give **3**, and (ii) the samarium iodide induced reductive opening of the oxygen bridge in **4** to yield hydroxyketone **5**, as key-steps.

Since the discovery of 1 α ,25-dihydroxyvitamin D₃ as the active metabolite of the vitamin many synthetic efforts have been produced in this area.^{1,2} The more recent discovery of a much expanded spectrum of biological activities has further initiated an impressive search for analogues with potential therapeutic activity.^{3,4}

Basic strategies for the construction of the 1 α -hydroxylated vitamin D skeleton include : (i) partial syntheses involving the introduction of the 1 α -hydroxy group onto the vitamin D structure or the photochemical cleavage of an adequately functionalized 7-dehydrocholesterol derivative, followed by the spontaneous rearrangement of the resulting previtamin triene to vitamin D;⁵ (ii) total syntheses which mostly involve the coupling of an A-ring precursor with the CD-ring part of the vitamin D skeleton, hence offering the advantages of convergency and flexibility.⁶ Most popular among the latter is the Wittig-Horner reaction of the lithium carbanion derived from phosphine oxide **I** (equation 1),⁷ based upon the original approach of Lythgoe and co-workers.⁸ In this paper we wish to describe a novel stereoselective synthesis of (\pm)-**10**, a known precursor of **I**.⁹

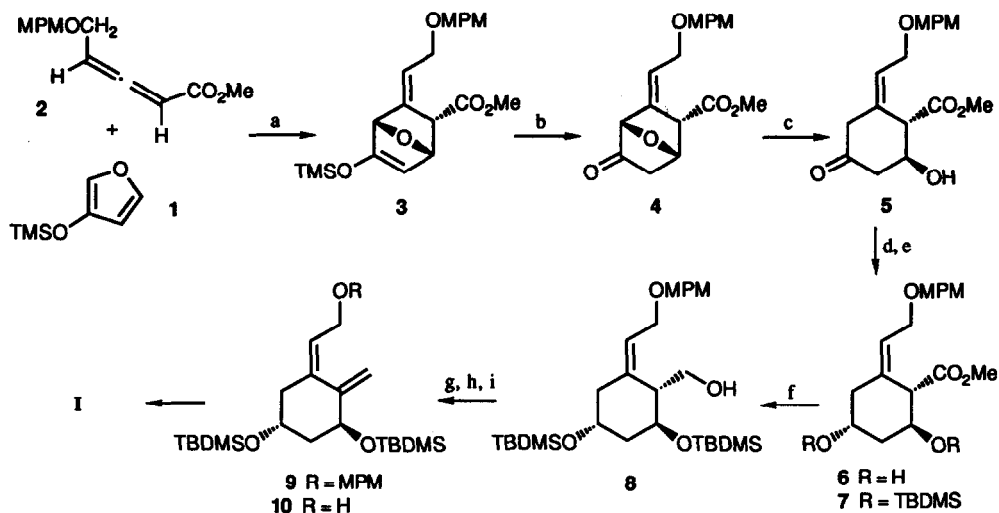


Central in the strategy stands the stereoselective cycloaddition between a 1,3-disubstituted allene and a 1,3-dioxydiene to give an adequately functionalized, full A-ring skeleton possessing all the required stereochemical features, in particular the (*Z*)-configuration at $\Delta^{5,6}$ and the 1 α -configuration at C-1.



As shown in scheme 1 obtention of the latter configuration can involve the combination of either (a) the (M)-allene with the (Z)-diene to give **IIIa**, or (b) the (P)-allene with the (E)-diene to give **IIIe**. In both cases the shown stereochemical features, i.e., the relative configuration at stereocenters C-1 and C-10, and the (Z)-configuration at the $\Delta^{5,6}$ -bond, result from the preferred endo-addition to the least hindered side of the allene.¹⁰ In both cases we note that the absolute configuration at C-1 is determined by the chirality of the allene (via C-10). The further obtention of the *trans*-relation between the protected hydroxy groups in **10** will result from an internal assisted hydride reduction requiring the axial orientation of the hydroxy group at C-1 (vide infra); this orientation, however, would be available in **IIIa**, but not in **IIIe**, since it was anticipated that the severe allylic strain caused by the substituent at C-6 would dominate the ring conformation by enforcing the methoxycarbonyl group at C-10 in the axial position.¹¹ Following the above consideration a diene of the (Z)-type is thus required. Because of our longstanding interest in the use of furan as a diene in (intramolecular) Diels-Alder reactions,¹² we decided to investigate the potential of 3-trimethylsilyloxyfuran in the present context.^{13,14} The resulting route is shown in scheme 2.

The cycloaddition between the known 3-trimethylsilyloxyfuran (**1**)¹⁵ and an excess of the p-methoxyphenylmethyl protected methyl 5-hydroxypenta-2,3-dienoate (**2**)¹⁶ leads smoothly (4 days, r.t., ether) to adduct **3** and, after methanol work-up, to bicyclic ketone **4** in 53 % isolated yield.¹⁷ Only one diastereomer was detected, the moderate yield being due to the slow decomposition of the allene. The shown configuration in **4** is substantiated by the vicinal coupling $J(1,10) = 5.7$ Hz.^{10,18} The subsequent crucial reductive opening of the oxygen bridge in **4** was realized in good yield (56 % isolated **5** with 22 % of starting material)¹⁷ with samarium(II)iodide (THF MeOH, -78°C),¹⁹ followed by careful low temperature acid quenching (2 equivs of acetic acid in THF, -90°C). Under these conditions no double bond isomerization is observed. The *trans*-diaxial conformation of **5** follows from the vicinal coupling $J(1,10) = 3.8$ Hz). The subsequent reduction of the hydroxy ketone **5** with aluminum hydride (THF, -78°C) led stereoselectively to the *trans*-diol **6** (62 % yield together with some starting material)¹⁷ presumably via prior complexation with the axially oriented 1 α -hydroxy group.²⁰ The equatorial orientation of the 3-hydroxy group follows from ¹H NMR analysis.¹⁷



(a) Et₂O, r.t., 4 days; (b) MeOH, r.t., 10 min (53 %); (c) SmI₂, THF, -78°C, 10 min; CH₃COOH, THF, -90°C (56 %); (d) AlH₃, THF, -78°C, 70 min (62 %); (e) t.butyldimethylsilyltriflate, 2,6-lutidine, CH₂Cl₂, r.t., 105 min (71 %); (f) LAH, THF, r.t., 30 min (70 %); (g) *o*-nitrophenyl selenocyanate, n.Bu₃P, THF, r.t., 2 h (55 %); (h) H₂O₂, CH₂Cl₂, r.t., 18 h (70 %); (i) DDQ, H₂O, CH₂Cl₂, r.t. (50 %).

Scheme 2

After protection of the diol (t-butyldimethylsilyltriflate, dichloromethane; 71 % yield),²¹ the ester in 7 was reduced (lithium aluminum hydride, THF, r.t.; 70 % yield), and the resulting primary alcohol 8 dehydrated ((i) : *o*-nitrophenyl selenocyanate; (ii) : hydrogen peroxide)²² to 9. Eventual deprotection (DDQ, dichloromethane-water)²³ led to the alcohol 10, the structure of which was fully confirmed by spectral comparison with authentic (+)-10.²⁴

The practical importance of the above route mainly resides in the possibility of including modifications which could be useful for the preparation of 1 α -hydroxylated A-ring analogues. Obviously, a further requirement is the elaboration of the enantioselective version which is presently under development.²⁵

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 17. Relevant ¹H NMR data : **4** (CDCl₃) : 7.22 (2H, d : 8.6 Hz), 6.88 (2H, d : 8.6 Hz), 5.95 (1H, m), 5.14 (1H, t : 5.7 Hz), 4.59 (1H, s), 4.40 (1H, d : 11.5 Hz), 4.36 (1H, d : 11.5 Hz), 4.01 (1H, ddd : 1.4, 4.3, 14.0 Hz), 3.93 (1H, ddd : 1.6, 5.8, 14.0 Hz), 3.93-3.87 (1H, m), 3.81 (3H, s), 3.63 (3H, s), 2.46 (1H, ddt : 1.3, 5.7, 17.7 Hz), 2.26 (1H, d : 17.7 Hz); **5** (CDCl₃) : 7.24 (2H, d : 8.7 Hz), 6.88 (2H, d : 8.7 Hz), 5.84 (1H, ddd : 2.3, 6.3, 7.0 Hz), 4.62 (1H, m), 4.45 (3H, s), 4.10 (1H, ddd : 1.7, 7.0, 11.8 Hz), 3.94 (1H, ddd : 1.5, 6.3, 11.8 Hz), 3.91 (1H, d : 3.8 Hz), 3.81 (3H, s), 3.74 (3H, s), 3.37 (1H, d : 16.0 Hz), 3.06 (1H, d : 16.0 Hz), 2.84 (1H, dd : 3.5, 16.8 Hz), 2.56 (1H, bd : 16.8 Hz); **6** (CDCl₃) : 7.25 (2H, d : 8.6 Hz), 6.88 (2H, d : 8.6 Hz), 5.88 (1H, ddd : 1.6, 6.4, 7.1 Hz), 4.49 (1H, m), 4.45 (2H, s), 4.06 (1H, dd : 7.1, 11.7 Hz), 3.97 (1H, dddd : 4.7, 6.1, 12.5, 12.7 Hz), 3.91 (1H, ddd, 1.0, 6.4, 11.7 Hz), 3.80 (3H, s), 3.68 (3H, s), 3.64 (1H, d : 2.6 Hz), 2.55 (1H, dd : 4.7, 12.5 Hz), 2.15 (1H, brd : 12.7 Hz), 1.73 (1H, dt : 2.5, 12.7 Hz); **10** (CDCl₃) : 5.54 (1H, br t : 6.3 Hz), 5.17 (1H, br s), 4.77 (1H, br s), 4.41 (1H, br t : 6.0 Hz), 4.22-4.15 (3H, m), 2.41 (1H, br dd : 4.0, 12.0 Hz), 1.84-1.80 (2H, m), 0.88 (9H, s), 0.87 (9H, s) 0.09 (6H, s), 0.04 (6H, s) ppm.
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